

Part of this material was deproteinised by partition between phenol and water², and the mucilage (8% of the air-dried stipules) recovered from the aqueous phase by dialysis and freeze-drying.

Fractionation of the mucilage by chromatography on a column of DEAE-Trisacryl M, using gradient elution with aqueous sodium chloride, gave a minor and a major acidic fraction, eluted with ~ 0.35 and ~ 0.60 M sodium chloride, respectively. The minor component, which on hydrolysis with acid yielded rhamnose, arabinose, xylose, mannose, glucose, and galactose, was not further investigated. The major component, $[\alpha]_D^{20} + 96$, was eluted as a single peak, with the void volume, on chromatography on a column of Bio-Gel P-100.

Hydrolysis of the major polysaccharide (PS) gave L-rhamnose as the only neutral sugar. When the PS was carboxyl-reduced³, sugar analysis of the product gave L-rhamnose, 4-O-methyl-D-glucose, and D-galactose in the proportions 1:1:2. The same result was obtained on sugar analysis of material prepared by methanolysis of the PS, followed by acetylation, and reduction with lithium borodeuteride in tetrahydrofuran. The absolute configurations of rhamnose and galactose were determined as devised by Gerwig *et al.*⁴. The 4-O-methylglucose was assumed to be D, as the corresponding uronic acid, 4-O-methyl-D-glucuronic acid, is a common component of plant polysaccharides.

Methylation analysis of the PS, with carboxyl-reduction of the methylated product, and of the carboxyl-reduced PS, gave the sugars listed in Table I, columns A and B. The results indicate that the PS is composed of L-rhamnosyl residues linked through O-2 and O-4, D-galactosyluronic acid residues linked through O-4, and terminal 4-O-methyl-D-glucosyluronic acid groups in the proportions $\sim 1:2:1$.

TABLE I

Methylation analyses of the rhamnogalacturonan and the tetrasaccharide I obtained on solvolysis with liquid hydrogen fluoride^a

Sugar ^c	T(1) ^b	T(2) ^b	Mole%		
			A	B	C
2,3,4,6-Glc	1.00	-		25	
3-Rha	1.17	1.15	30	25	30
2,3,6-Gal	1.28	-		50	
2,3,4-Glc	1.39	1.37	20		11
2,3,4-Gal	-	1.40			9
2,3-Gal	1.77	1.70	50		26
3,4-Gal	-	1.78			24

^a Key: A, PS carboxyl-reduced after methylation; B, carboxyl-reduced PS; C, tetrasaccharide, carboxyl-reduced after methylation. ^b 2,3-Rha = 2,3-di-O-methyl-L-rhamnose, *etc.* ^c Retention time of the corresponding alditol acetate, relative to 1,5-di-O-acetyl-2,3,4,6-tetra-O-methyl-D-glucitol on an HP 54 column. Temperature program: 190° (3 min) \rightarrow 250° at 3°/min. ^d Same as ^c but with the program: 175° (3 min) \rightarrow 250° at 3°/min.

The PS gave viscous solutions and poor n.m.r. spectra. The ^1H -n.m.r. spectrum showed, *inter alia*, signals for *O*-acetyl groups and methyl groups of L-rhamnosyl residues in the ratio $\sim 2:1$.

In order to obtain better spectra, part of the PS was *O*-deacetylated and hydrolysed with aqueous trifluoroacetic acid under mild conditions. The ^1H -n.m.r. spectrum (Fig. 1) of this product, which was still polymeric, contained several signals in the "anomeric region", some of which are given by H-5 of uronic acid residues. The ^{13}C -n.m.r. spectrum contained signals for anomeric carbons at δ 103.8, 100.5, 99.8, and 99.0. In agreement with this finding, the ^1H -n.m.r. spectrum (Fig. 2) of the *O*-deacetylated and carboxyl-reduced PS contained signals for anomeric protons at δ 5.21 (not resolved, $\nu_{1/2} \sim 1$ Hz, 1 H), 5.08 (not resolved, $\nu_{1/2} \sim 4$ Hz, 1 H), 4.93 (J 3.7 Hz, 1 H), and 4.67 ($J \sim 7$ Hz, 1 H). The ^{13}C -n.m.r. spectrum contained signals for four anomeric carbons at δ 104.5, 100.8 (2 C), and 100.3. These results indicate that all of the sugar residues are pyranosidic, that one of the uronic acid residues is β -linked, and that the three other sugar residues are α -linked.

In order to determine the sequence of the sugar residues, the *O*-deacetylated PS was treated with liquid hydrogen fluoride at -27° , conditions under which only the rhamnosyl linkages should be cleaved⁵. The main product was isolated by chromatography on Bio-Gel P-2, and was eluted in the tetrasaccharide region. This substance, on sugar analysis with carboxyl reduction, yielded rhamnose, 4-*O*-methylglucose, and galactose in the proportions 1:1:2. Attempted reduction of the tetrasaccharide with sodium borohydride showed that it was non-reducing. Methylation analysis, with carboxyl reduction of the methylated product, gave the sugars listed in Table I, column C. The formation of 3,4-di-*O*-methyl-D-galactose and the fact that the tetrasaccharide is

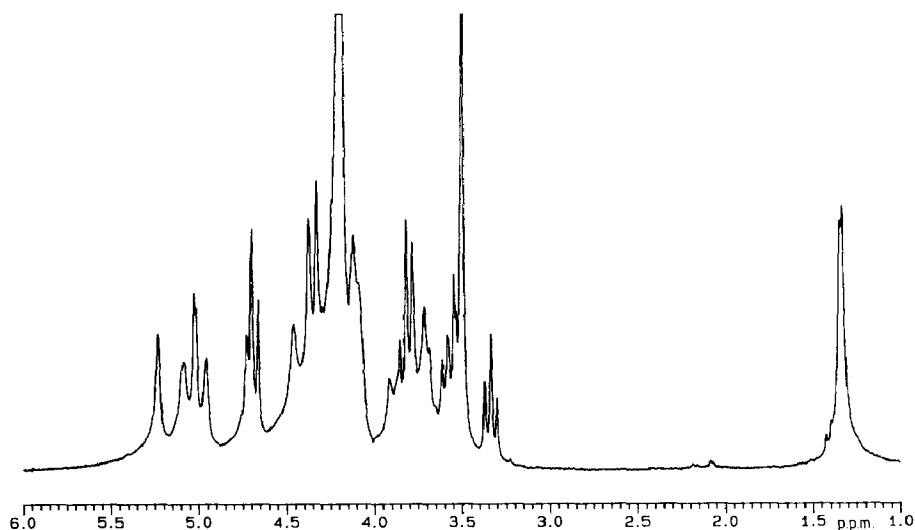


Fig. 1. ^1H -N.m.r. spectrum of the *O*-deacetylated, partially depolymerised PS.

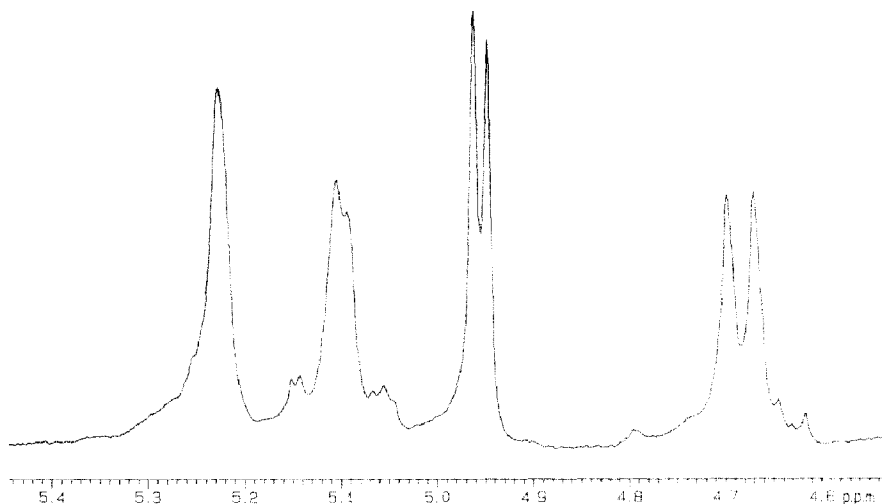
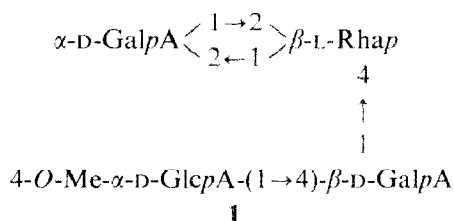


Fig. 2. Low-field part of the ^1H -n.m.r. spectrum of the carboxyl-reduced PS.

non-reducing demonstrate that the sequence of sugar residues is that given in structure **1**, with a dianhydride component.



In structure **1**, the L-rhamnosyl residue has to be β -linked for steric reasons, which is also evident from the n.m.r. spectra. The ^1H -n.m.r. spectrum (Fig. 3) thus contains signals for four anomeric protons at δ 5.25 (J 3.6 Hz, 1 H), 5.02 (J 3.7 Hz, 1 H), 4.85 (J \sim 1 Hz, 1 H), and 4.79 (J 7.9 Hz, 1 H), and that at δ 4.85 is given by the β -L-rhamnopyranosyl residue. The signals for the anomeric carbons appear at δ 104.1, 100.7, 95.5, and 92.1. The torsional angles at the glycosidic linkages in the dianhydride moiety differ from those for ordinary oligosaccharides, which may explain the low values of some of the chemical shifts observed. The signals given by all protons of the four sugar residues in the tetrasaccharide were assigned from COSY spectra (Table II). The chemical shifts of the resonances of the dianhydride moiety coincide well with those given by Komalavilas and Mort⁷. It is evident, from the chemical shifts and coupling constants, that the terminal 4-O-Me-D-GlcpA ($J_{1,2}$ 3.7, $J_{3,4}$ 8.8 Hz) is α -linked and that one of the two D-GalpA residues ($J_{1,2}$ 7.3, $J_{3,4}$ 3.7 Hz) is β -linked. That this is the residue in the side chain was demonstrated by a NOESY spectrum which, *inter alia*, showed

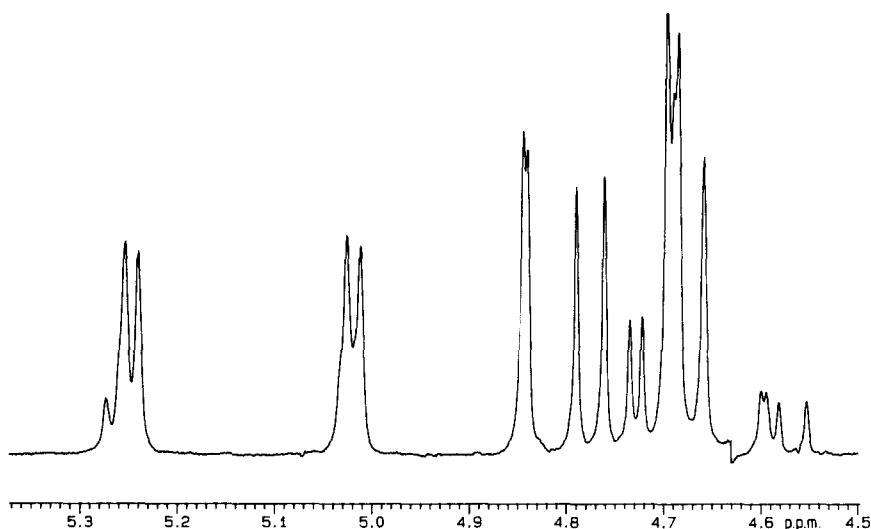


Fig. 3. Low-field part of the ^1H -n.m.r. spectrum of the tetrasaccharide.

TABLE II

Chemical shifts in the ^1H -n.m.r. spectrum of the tetrasaccharide 1

Sugar residue	Chemical shift ^a (δ)					
	H-1	H-2	H-3	H-4	H-5	H-6
$\rightarrow 2$)- α -D-GalpA-(1 \rightarrow	5.25 (3.1)	3.88	4.71	4.35	4.71	
α -D-GlcpA-4Me-(1 \rightarrow	5.02 (3.7)	3.52	3.80 (8.8)	3.33	4.70	
$\rightarrow 4$)- β -L-Rhap-(1 \rightarrow	4.85 (1.5)	4.21	3.96	3.72	3.45	1.41
2 ↑						
$\rightarrow 4$)- β -D-GalpA-(1 \rightarrow	4.79 (7.3)	3.57	3.81 (3.7)	4.35	4.35	

^a J values (Hz) in parentheses.

n.O.e. contacts between H-1 of 4-*O*-Me- α -D-GlcpA and H-4 of β -D-GalpA, and between H-1 of β -D-GalpA and H-4 of β -L-Rhap.

The formation of 2,3,4-tri-*O*-methyl-D-galactose in the methylation analysis of the tetrasaccharide may be due to partial elimination of the terminal 4-*O*-methyl-D-glucosyluronic acid residue during the methylation. The formation of tetrasaccharide 1 on solvolysis with liquid hydrogen fluoride nevertheless demonstrates that the PS is, at least mainly, composed of tetrasaccharide repeating-units having the structure 2. Rhamnogalacturonans with the same linear backbone as 2 are common in plants⁶, but they generally have complicated and non-regular structures. The finding of a polysaccharide belonging to this group, but with a simple, essentially regular structure, was therefore unexpected.

Part of the *O*-deacetylated material was treated with 0.5M trifluoroacetic acid at 70° for 2 h, followed by chromatography on a column of Bio-Gel P-2. The main fraction, eluted in the void volume, was used for n.m.r. studies.

Carboxyl reduction of the polysaccharide. — This was performed as devised by Taylor *et al.*³, and the procedure was repeated once.

Sugar analysis. — The material (~1 mg) in dry methanol (1.5 mL) to which acetyl chloride (200 μ L) had been added, in a Teflon-lined screw-cap vial, was kept at 80° for 16 h. After cooling, the contents were neutralised with silver carbonate, centrifuged, concentrated, and acetylated by treatment with acetic anhydride–pyridine. A solution of the acetylated material in dry tetrahydrofuran (2.5 mL) containing lithium borohydride (25 mg) was kept at 70° for 2 h. After conventional work-up, the product was hydrolysed by treatment with aqueous 2M trifluoroacetic acid at 120° for 2 h, and the products were reduced with sodium borohydride, acetylated, and analysed by g.l.c.

Methylation analysis. — This was performed as previously described^{10,11}. Methylated products containing uronic acid residues were carboxyl-reduced, using lithium borohydride in tetrahydrofuran, before they were hydrolysed.

Solvolysis with liquid hydrogen fluoride. — The *O*-deacetylated PS (40 mg) was dissolved in anhydrous hydrogen fluoride (2 mL) and kept at –27° for 30 min. Diethyl ether (5 mL) was added and the hydrogen fluoride–ether complex evaporated under reduced pressure at room temperature. The product was fractionated on a column of Bio-Gel P-2, using 50mM acetate buffer of pH 5.2 as irrigant. The major component (23 mg) was isolated by treatment with Dowex 50 (H⁺) resin and freeze-drying.

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